

Lecture #5 Time-independent

Schrodinger Equation

Last time

$$\hat{p} = -i\hbar \frac{\partial}{\partial x}$$

$$\hat{x} = x$$

Gaussian Wave Packet

$$\Psi = A e^{-\frac{\lambda}{2}x^2}$$

You calculate

$$\langle x \rangle = 0$$

$$\langle x^2 \rangle = 1/2 \lambda$$

Symmetry

Use $\int_{-\infty}^{\infty} dx e^{-x^2} = \sqrt{\pi}$

$$\begin{aligned} \langle p \rangle &= \int dx A^* e^{-\frac{\lambda}{2}x^2} -i\hbar \frac{\partial}{\partial x} A e^{-\frac{\lambda}{2}x^2} \\ &= A^* A i\hbar \lambda \int_{-\infty}^{\infty} dx e^{-\frac{\lambda}{2}x^2} \times e^{-\frac{\lambda}{2}x^2} \equiv 0 \end{aligned}$$

\downarrow odd

$$\hat{p}^2 = -\hbar^2 \frac{\partial^2}{\partial x^2}$$

$$\begin{aligned} \hat{p}^2 \Psi &= -\hbar^2 \frac{\partial}{\partial x} A(-\lambda x e^{-\frac{\lambda}{2}x^2}) \\ &= +\hbar^2 A \lambda \left[e^{-\frac{\lambda}{2}x^2} - \lambda x^2 e^{-\frac{\lambda}{2}x^2} \right] \end{aligned}$$

$$= \hbar^2 [\lambda - \lambda^2 x^2] \Psi$$

$$\begin{aligned} \langle p^2 \rangle &= \int_{-\infty}^{\infty} dx \Psi^* \hat{p}^2 \Psi \\ &= \hbar^2 \int_{-\infty}^{\infty} dx \Psi^* [\lambda - \lambda^2 x^2] \Psi \\ &= \hbar^2 [\lambda - \lambda^2 \langle x^2 \rangle] \end{aligned}$$

$$\langle p^2 \rangle = \hbar^2 \left[\lambda - \lambda^2 \frac{1}{2\lambda} \right] = \frac{\hbar^2 \lambda}{2}$$

$$\langle \Delta x^2 \rangle^{1/2} = [\langle x^2 \rangle - \langle x \rangle^2]^{1/2} = \left[\frac{1}{2\lambda} - 0 \right]^{1/2}$$

$$\Delta x = \frac{1}{(2\lambda)^{1/2}}$$

$$\Delta p \equiv \langle (\Delta p)^2 \rangle^{1/2} = [\langle p^2 \rangle - \langle p \rangle^2]^{1/2}$$

$$\Delta p = \left[\hbar^2 \frac{\lambda}{2} - 0 \right]^{1/2} = \hbar \left(\frac{\lambda}{2} \right)^{1/2}$$

and finally $\Delta p \Delta x = \hbar \left[\frac{\lambda}{2} \frac{1}{2\lambda} \right]^{1/2}$

$$\Delta p \Delta x = \frac{\hbar}{2}$$

(a) Independent of λ

(b) You can trade: a large $\lambda \rightarrow$ small Δx
but also must then have large Δp
Small $\lambda \rightarrow$ small Δp but large Δx

(c) Gaussian packet has minimum $\Delta p \Delta x$
other wave functions can have large uncert.
Thus

Heisenberg Uncertainty Principle

$$\Delta p \Delta x \geq \frac{\hbar}{2}$$

(1) Can determine Δx or Δp to arbitrary accuracy. Just not both

(2) One way to think about it. Need at least one photon to bounce off an electron to see it.

Wavelength λ of photon restricts

$$\Delta x \sim \lambda$$

Need to use short wave length radiation to make good Δx measurement.

But $p = h/\lambda = 2\pi\hbar/\lambda$
(de Broglie formula)
is momentum of photon. In bouncing off the electron the recoil will impart a Δp to the electron.

$$\Delta p \sim p \sim h/\lambda$$

Thus clearly Heisenberg U. principle is a limitation on our information about a particle.

However it goes much deeper. It is built into the formalism of quantum mechanics. You can not write down a wave function with better $\Delta p \Delta x$.

People go so far as to say extra information does not exist (Orthodox or Copenhagen interpretation). The particle did not have a position when I measured its momentum.

Chap. 2 Time-Independent Schrodinger Equation

Want to solve Schrodinger equation.

$$i\hbar \frac{\partial \Psi(x,t)}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x,t)}{\partial x^2} + V(x) \Psi(x,t)$$

Complicated partial dif. equation Ψ function both of x, t .

1st thing to try because it is simple is separation of variables, can we write

$$\Psi(x,t) = \psi(x) f(t) \quad \textcircled{A}$$

In general this will not be true. However even then we may be able to use a sum

$$\Psi(x,t) = \sum \psi_n(x) f_n(t)$$

But if it is true it makes life much easier.

Plug \textcircled{A} into S. eq.

$$\psi(x) \left[i\hbar \frac{df(t)}{dt} \right] = f(t) \left(-\frac{\hbar^2}{2m} \frac{d^2 \psi(x)}{dx^2} + V(x) \psi(x) \right)$$

Divide through by ψf

$$i\hbar \frac{1}{f} \frac{df}{dt} = \frac{1}{\psi} \left[-\frac{\hbar^2}{2m} \frac{d^2 \psi}{dx^2} + V(x) \psi(x) \right]$$

Left hand side only a function of t
 Right hand side " " " "
 if true for all t, x they must be equal to a constant.

Call the constant E (it turns out to be the particles energy but we will see that later)

$$i\hbar \frac{1}{F(t)} \frac{dF}{dt} = E \quad \textcircled{B}$$

$$-\frac{\hbar^2}{2m} \frac{1}{\psi} \frac{d^2\psi}{dx^2} + V = E$$

or

$$\boxed{-\frac{\hbar^2}{2m} \frac{d^2\psi(x)}{dx^2} + V(x)\psi(x) = E\psi(x)} \quad \textcircled{C}$$

This is time-indep. Schrodinger equation

Can think of $\hat{p} = i\hbar \frac{\partial}{\partial x}$

$$\left[\frac{\hat{p}^2}{2m} + V(x) \right] \psi(x) = E\psi(x)$$

Compared to classical mechanics -

$$E = T + V = \frac{p^2}{2m} + V$$

Solve \textcircled{B}

$$\int \frac{dF}{F} = \frac{-i}{\hbar} E \int dt$$

$$\ln F = \frac{-iE}{\hbar} t + C$$

\Rightarrow

$$\boxed{F = e^{-\frac{iE}{\hbar} t}}$$

\textcircled{A} IF we can separate variables then

$$\Psi(x,t) = \Psi(x) e^{-i\frac{E}{\hbar}t}$$

is called a stationary state because
prob. density $\Psi^* \Psi$ is time
independent

$$\begin{aligned} \Psi^* \Psi(x,t) &= \Psi^*(x) e^{i\frac{E}{\hbar}t} \Psi(x) e^{-i\frac{E}{\hbar}t} \\ &= \Psi^*(x) \Psi(x) \quad \text{indep. of time} \end{aligned}$$

(B) Stationary states have definite energy

Operator corresponds to energy

$$\hat{H} = \frac{\hat{p}^2}{2m} + V$$

$$\langle \hat{H} \rangle = \langle E \rangle \quad \text{average energy}$$

Note $\hat{H} \Psi(x) = E \Psi(x)$

$$\begin{aligned} \text{So } \int dx \Psi^*(x) \hat{H} \Psi(x) &= \langle \hat{H} \rangle \\ &= \int dx \Psi^*(x) E \Psi(x) = E \int dx \Psi^*(x) \Psi(x) = E \end{aligned}$$

Let's calculate spread in energy

$$\langle (\Delta E)^2 \rangle = \langle \hat{H}^2 \rangle - \langle \hat{H} \rangle^2$$

$$\begin{aligned} \langle \hat{H}^2 \rangle &= \int dx \Psi^* \hat{H} \hat{H} \Psi \\ &= \int dx \Psi^* \hat{H} (E \Psi) = E \int dx \Psi^* \hat{H} \Psi \end{aligned}$$

$$\langle (\Delta E)^2 \rangle = E^2 - E^2 = 0 \Rightarrow \Delta E \equiv 0$$