

3/5/01

# Lec. 23 Hydrogen Atom

$$V(r) = -\frac{e^2}{4\pi\epsilon_0} \frac{1}{r}$$

Radial Equation

$$-\frac{\hbar^2}{2m} \frac{d^2 u}{dr^2} + \left[ -\frac{e^2}{4\pi\epsilon_0} \frac{1}{r} + \frac{\hbar^2}{2m} \frac{l(l+1)}{r^2} \right] u = E u$$

$$\Psi(r, \theta, \phi) = \frac{u(r)}{r} Y_l^m(\theta, \phi)$$

$$k = \frac{\sqrt{-2mE}}{\hbar}$$

lock for bound states  $E < 0$

Note also scattering states  $E > 0$

Let  $\rho \equiv kr$        $\rho_0 \equiv \frac{me^2}{2\pi\epsilon_0 \hbar^2 k}$

$$\boxed{\frac{d^2 u}{d\rho^2} = \left[ 1 - \frac{\rho_0}{\rho} + \frac{l(l+1)}{\rho^2} \right] u}$$

- Procedure
- (1) Find large  $\rho$  and small  $\rho$  behavior of  $u(\rho)$
  - (2) Expand remainder in a power series  $\sum a_i \rho^i$
  - (3) Require series to terminate so that  $u(\rho)$  is normalizable  
 $\int_0^\infty |u|^2 = 1$
  - (4) This gives allowed energies

For large  $\rho$   $\frac{d^2 u}{d\rho^2} \approx u \Rightarrow u = Ae^{-\rho} + Be^{\rho}$   
 choose  $e^{-\rho}$  since  $e^{\rho}$  blows up at large  $\rho$ .

For very small  $\rho$

$$\frac{d^2 u}{d\rho^2} \approx \frac{l(l+1)}{\rho^2} u$$

$u(\rho) = C \rho^{l+1} + D \rho^{-l}$  as  $\rho \rightarrow 0$   
but  $\rho^{-l}$  blows up so keep  $\rho^{l+1}$

Guess

$$u(\rho) = \rho^{l+1} e^{-\rho} v(\rho)$$

This defines remainder function  $v(\rho)$ .  
Plug into

$$\frac{d^2 u}{d\rho^2} = \left[ 1 - \frac{\rho_0}{\rho} + \frac{l(l+1)}{\rho^2} \right] u$$

$$\rho \frac{d^2 v}{d\rho^2} + 2(l+1-\rho) \frac{dv}{d\rho} + [\rho_0 - 2(l+1)] v = 0 \quad (*)$$

Assume a power series

$$v = \sum_{j=0}^{\infty} a_j \rho^j$$

$$\frac{dv}{d\rho} = \sum_j j a_j \rho^{j-1}$$

$$\frac{d^2 v}{d\rho^2} = \sum_j j(j-1) a_j \rho^{j-2}$$

put into equation (\*) and require each power of  $\rho$  vanish

$$a_{j+1} = \left\{ \frac{2(j+l+1) - \rho_0}{(j+1)(j+2l+2)} \right\} a_j$$

For wave function to be normalizable and  $\rightarrow 0$  as  $\rho \rightarrow \infty$  need series to terminate

$$a_{j_{\max}+1} = 0$$

let

$$n \equiv j_{\max} + l + 1$$

$$\Rightarrow \rho_0 = 2n = \frac{m e^2}{2\pi\epsilon_0 \hbar^2 \kappa}$$

$$E = -\frac{\hbar^2 \kappa^2}{2m} = -\frac{m e^4}{8\pi^2 \epsilon_0^2 \hbar^2 \rho_0^2}$$

$$E = -\frac{m}{2} \left( \frac{e^2}{4\pi\epsilon_0 \hbar} \right)^2 \frac{1}{n^2} = \frac{E_1}{n^2} \quad n=1,2,3,\dots$$

$$\kappa = \frac{m e^2}{4\pi\epsilon_0 \hbar^2} \frac{1}{n} = \frac{1}{a n} \quad \text{Bohr Formula}$$

$$a \equiv \left( \frac{4\pi\epsilon_0 \hbar^2}{m e^2} \right) = 0.529 \times 10^{-10} \text{ m}$$

Bohr radius

Note  $\rho = r/a n$

$$\alpha = \text{Fine structure constant} \equiv \frac{e^2}{4\pi\epsilon_0 \hbar c}$$

$$\approx \frac{1}{137.036}$$

$$E = -\frac{m c^2}{2} \alpha^2 \frac{1}{n^2}$$

$$m c^2 = 0.511 \text{ MeV}$$

rest mass of electron

$$\hbar c = 197.33 \text{ MeV-fm} = 1973.3 \text{ eV-Å}$$

$$\Psi_{nlm}(r, \theta, \phi) = R_{nl}(r) Y_l^m(\theta, \phi)$$

$$R_{nl} = \frac{1}{r} \rho^{l+1} e^{-\rho} v(\rho)$$

$$a_{j+1} = \frac{2(j+l+1-n)}{(j+1)(j+2l+2)} a_j$$

$$E_1 = -\frac{m}{2\hbar^2} \left( \frac{e^2}{4\pi\epsilon_0} \right)^2 = -\frac{mc^2}{2} \alpha^2$$

$$= -13.6 \text{ eV}$$

Ground state  $n=1 = j_{\max} + l + 1$   
 let  $l = j_{\max} = 0$

$$a_1 = \frac{2(0+0+1-1)}{1 \cdot 3} a_0 = 0$$

$$V(\rho) = a_0 \quad a_2 = a_3 = \dots = 0$$

$$R_{10} = \frac{a_0}{a} e^{-r/a}$$

Normalize

$$\int_0^{\infty} |R_{10}|^2 r^2 dr = 1$$

$$R_{10}(r) = \frac{2}{a^{3/2}} \frac{1}{a} e^{-r/a}$$

$$Y_0^0 = \frac{1}{\sqrt{4\pi}}$$

$$\Psi_{100} = R_{10} Y_0^0$$

$$\boxed{|\Psi_{100}(r, \theta, \phi) = \frac{1}{(\pi a^3)^{1/2}} e^{-r/a}}$$