

4/6/01

# Lecture 32 Addition of Ang. Mom. cont.

Add angular momentum  $l$  (orbital) to spin  $\frac{1}{2}$  to get total angular momentum  $J$

$$|JM\rangle = \sum_{m_l, m_s} C_{m_l, m_s}^{l, \frac{1}{2}, J} Y_{l, m_l} X_{m_s}$$

IF  $M$  is the maximum possible value

$$M = l + \frac{1}{2}$$

Only one way to make it

$$m_l = l \quad m_s = +\frac{1}{2}$$

also add  $s = \frac{1}{2}$  to  $l$  allowed  $J$  values

$$|l - \frac{1}{2}| \leq J \leq l + \frac{1}{2}$$

and  $J \geq |M| \Rightarrow J = M = l + \frac{1}{2}$

$$|l + \frac{1}{2}, l + \frac{1}{2}\rangle = Y_{l, l} X_{+\frac{1}{2}}$$

so

$$C_{l, \frac{1}{2}, l + \frac{1}{2}}^{l, \frac{1}{2}, J} = 1$$

Now apply  $J_- = L_- + S_-$

$$J_- |JM\rangle = A_J^M |J, M-1\rangle$$

From problem (4.19)

$$A_J^M = \hbar \sqrt{J(J+1) - M(M-1)}$$

$$J = M = l + \frac{1}{2}$$

$$A_J^M = \hbar \sqrt{(l + \frac{1}{2})(l + \frac{3}{2}) - (l + \frac{1}{2})(l - \frac{1}{2})} = \hbar \sqrt{2l + 1}$$

$$A_J^M |J, J-1\rangle = A_l^l Y_l^{l-1} X_{1/2} + A_s^{1/2} Y_l^l X_{-1/2}$$

$$A_l^l = \hbar \sqrt{l(l+1) - l(l-1)} = \hbar \sqrt{2l}$$

$$A_s^{1/2} = \hbar \sqrt{\frac{1}{2}(3/2) - \frac{1}{2}(-1/2)} = \hbar$$

$$\hbar \sqrt{2l+1} |J, J-1\rangle = \hbar (\sqrt{2l} Y_l^{l-1} X_{1/2} + Y_l^l X_{-1/2})$$

So

$$|J, J-1\rangle = \sqrt{\frac{2l}{2l+1}} Y_l^{l-1} X_{1/2} + \frac{1}{\sqrt{2l+1}} Y_l^l X_{-1/2}$$

$$\Rightarrow \begin{matrix} \langle l, 1/2, l+1/2 \\ l-1, 1/2, l-1/2 \end{matrix} = \sqrt{\frac{2l}{2l+1}}, \quad \begin{matrix} \langle l, 1/2, l+1/2 \\ l, -1/2, l-1/2 \end{matrix} = \frac{1}{\sqrt{2l+1}}$$

State  $|J, J-1\rangle, Y_l^{l-1} X_{1/2}, \dots$  are normalized

$$\Rightarrow \sum_{m_l, m_s} \left| \langle l, \frac{1}{2}, l+1/2 \mid m_l, m_s, m_l+m_s \rangle \right|^2 = 1$$

$$= \frac{2l}{2l+1} + \frac{1}{2l+1} = 1 \quad \checkmark$$

## Identical Particles

Start reading chapter 5

Two particle system

$$\Psi(\vec{r}_1, \vec{r}_2; t)$$

$$i\hbar \frac{\partial \Psi}{\partial t} = \hat{H} \Psi$$

$$\hat{H} = -\frac{\hbar^2}{2m_1} \nabla_1^2 - \frac{\hbar^2}{2m} \nabla_2^2 + V(\vec{r}_1, \vec{r}_2)$$

Normalization

$$P(r_1, r_2) = |\Psi(\vec{r}_1, \vec{r}_2, t)|^2 d^3 r_1 d^3 r_2$$

is simultaneous prob. to find 1 within  $d^3 r_1$  at  $\vec{r}_1$  and within  $d^3 r_2$  of  $\vec{r}_2$  at same time to find 2

$$\int \int |\Psi(\vec{r}_1, \vec{r}_2, t)|^2 d^3 r_1 d^3 r_2 = 1$$

Six dimensional integral

Time independent Schrodinger eq. as before

$$\Psi(r_1, r_2, t) = \psi(\vec{r}_1, \vec{r}_2) e^{-iEt/\hbar}$$

$E$  = total energy of system

$$\hat{H} \psi(\vec{r}_1, \vec{r}_2) = E \psi(\vec{r}_1, \vec{r}_2)$$

$$\hat{H} = -\frac{\hbar^2}{2m_1} \nabla_1^2 - \frac{\hbar^2}{2m_2} \nabla_2^2 + V(\vec{r}_1, \vec{r}_2)$$

In general expect complex correlations between particles. For a two electron (He) atom because electrons repel each other expect

$$\psi \rightarrow 0 \quad \text{as} \quad |\vec{r}_1 - \vec{r}_2| \rightarrow 0$$

As a simple case, consider small interactions between particles.

Example He like Uranium ion.

Uranium (of 92) has  $Z=92$  (Nucleus has a charge of 92) If one has a Uranium atom that has been ionized 90 times so only two electrons are left

$$V(\vec{r}_1, \vec{r}_2) = -\frac{Ze^2}{4\pi\epsilon_0 r_1} - \frac{Ze^2}{4\pi\epsilon_0 r_2} + \frac{e^2}{4\pi\epsilon_0 |\vec{r}_1 - \vec{r}_2|}$$

$Z=92 \Rightarrow$  can neglect  $e-e$  repulsion term compared to the large attraction of each electron to the nucleus.

If a solution  $V(\vec{r}_1, \vec{r}_2) \approx V(r_1) V(r_2)$  look for

$$\psi(r_1, r_2) \approx \psi_a(r_1) \psi_b(r_2)$$

Separation of variables

This assumes one can tell the two electrons apart.

In classical mech. you can put a red dot on one of the electrons to tell it apart from the other.

In quantum mechanics all electrons are exactly identical. You would have to make the "red dot" out of some other particle but then you would no longer have just an electron.

Quantum mechanics can deal with particles that are indistinguishable even in principle.

Perhaps an additional extension of HUP I can't make a measurement that tells

which electron is which.

If I don't know that I was in state a and 2 state b, the other way around consider combination

$$\Psi_{\pm}(\vec{r}_1, \vec{r}_2) = A \left[ \Psi_a(\vec{r}_1) \Psi_b(\vec{r}_2) \pm \Psi_a(\vec{r}_2) \Psi_b(\vec{r}_1) \right]$$

Observed property

All integer spin particles, called bosons, are symmetric under interchange  $1 \leftrightarrow 2$

All  $\frac{1}{2}$  integer spin particles, called Fermions, are antisymmetric under  $1 \leftrightarrow 2$

He like Uranium atom

$$H_1 = -\frac{\hbar^2}{2m_1} \nabla_1^2 - \frac{Ze^2}{4\pi\epsilon_0 r_1}$$

$$H_2 = -\frac{\hbar^2}{2m_2} \nabla_2^2 - \frac{Ze^2}{4\pi\epsilon_0 r_2}$$

$$H_1 \Psi_a(r_1) = E_a \Psi_a(r_1)$$

$$H_2 \Psi_b(r_2) = E_b \Psi_b(r_2)$$

$$\Psi = A \left[ \Psi_a(r_1) \Psi_b(r_2) - \Psi_a(r_2) \Psi_b(r_1) \right]$$

$$E \approx E_a + E_b$$

because electrons are Fermions

Spin statistics theorem

all integer spin particles (photons  $(s=1)$ , mesons  $(s=0,1)$ ) are bosons

all half integer spin particles are  
Fermions  
electrons, quarks, ...

### Pauli exclusion principle

Two identical Fermion particles can not  
be in the same spatial and spin  
state.

$$A [\psi_a(r_1) \psi_a(r_2) - \psi_a(r_2) \psi_a(r_1)] \equiv 0$$